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A NEW SYNTHETIC ROUTE TO SULFINIC ACID ESTERS

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Dialkylsulfoxylates undergo reactions with alkylhalides and other alkylating agents to give sulfinic acid esters, which were characterized by IR-, ¹H-NMR- and mass-spectra.

$$R^{1}-I$$

$$-R-I$$

$$R^{1}-S-OR$$

$$R^{1}=CH_{3}, C_{2}H_{5}$$

$$+CH_{2}=CH-CH_{2}-Br$$

$$-R-Br$$

$$CH_{2}=CH-CH_{2}-S-OR$$

$$R=n-C_{3}H_{7}, i-C_{3}H_{7},$$

$$n-C_{4}H_{9}, i-C_{4}H_{9},$$

$$n-C_{5}H_{11}$$

$$+C_{6}H_{5}CH_{2}Br/CH_{3}NO_{2}$$

$$-R-Br$$

$$+MeOCOCH_{2}-Br$$

$$-R-Br$$

$$MeOC-CH_{2}-S-OR$$

The reaction conditions are discussed and a mechanism of the interaction is suggested. This type of reaction represents an alternative synthetic route to sulfinic acid esters.